



Phosphorus Chemistry and Sequestration in Soil

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Today's Presentation

- **Soil Chemical Processes Important to P sequestration and solubility / mobility / runoff / leaching**
- **Soil Testing of “labile P”:
Crop production vs environmental soil testing**
- **Assessment Tools to predict Risk of P transport (runoff/leaching) in Ohio**

Soil Phosphorus Pools

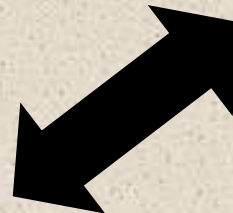
Sequestered P
very insoluble forms of P
insoluble P minerals
strongly adsorbed P

“Labile P”
P in Equilibrium with
soil solution P
Adsorbed (bound) P
soluble P minerals

Quantity

Intensity

Soil solution P
dissolved P
bioavailable P



Q-I is a 2 way street
replaces losses of dissolved P
removes additions of soluble P

Sequestered >> Quantity >> Intensity

Chemical Fate of Introduced Soluble P

Fertilizer / Manure / Biosolids Additions



Soluble
 H_2PO_4^-

**Adsorption (Binding) onto soil oxide clays
(Fe, Al, Mn oxides)**

**Precipitation (Fe, Ca, Al phosphates)
formation and transformation of P minerals**

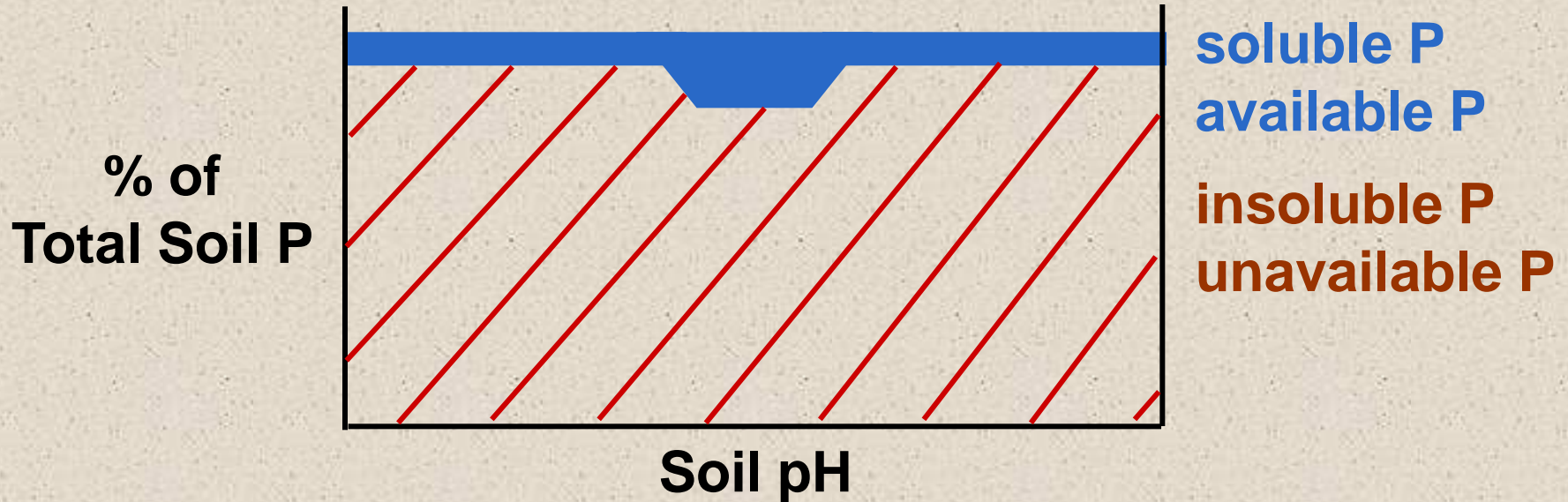
**These two mechanisms control labile P in
equilibrium with soil solution (QI)**

Precipitation / Dissolution of P Minerals

Solubility of P-minerals limits P dissolution and availability

Fe, Al phosphates at low pH (<5)

Ca phosphates at high pH (5 - 8.5)

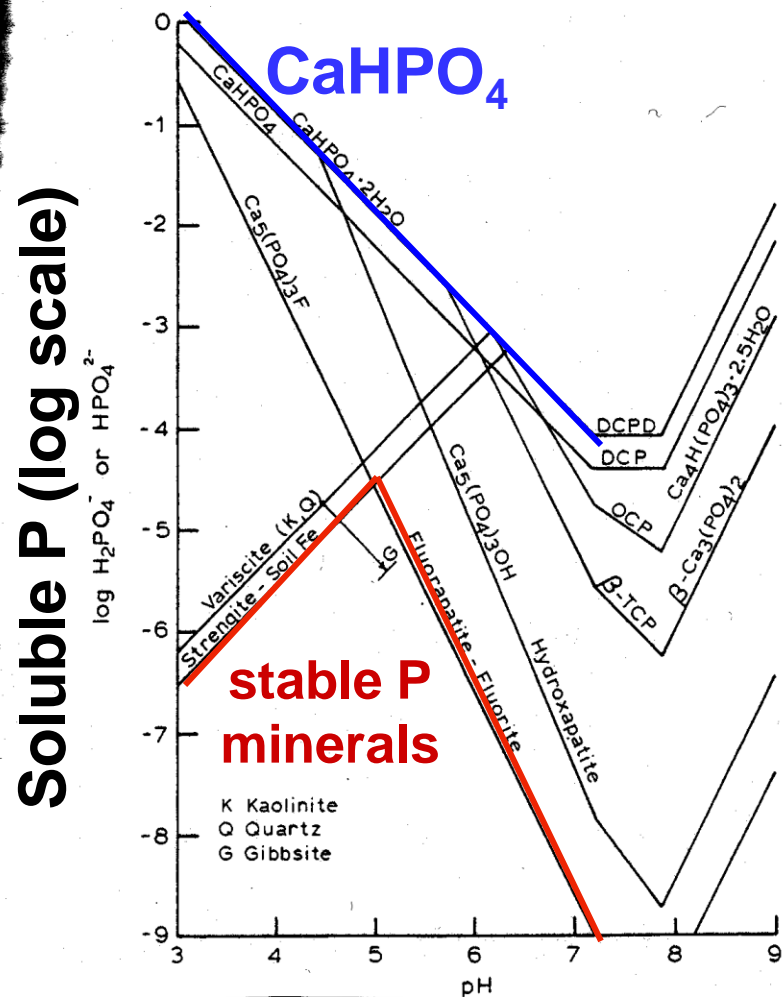


Very insoluble at most pH
maximum solubility at pH 5 to 7

plant available P can be very low in unfertilized soil
plant growth and **crop yield** can be limited by P

Does P fertilizer increase P solubility?

SOLUBILITY OF CALCIUM PHOSPHATES



Fertilizer P becomes CaHPO_4 in soil

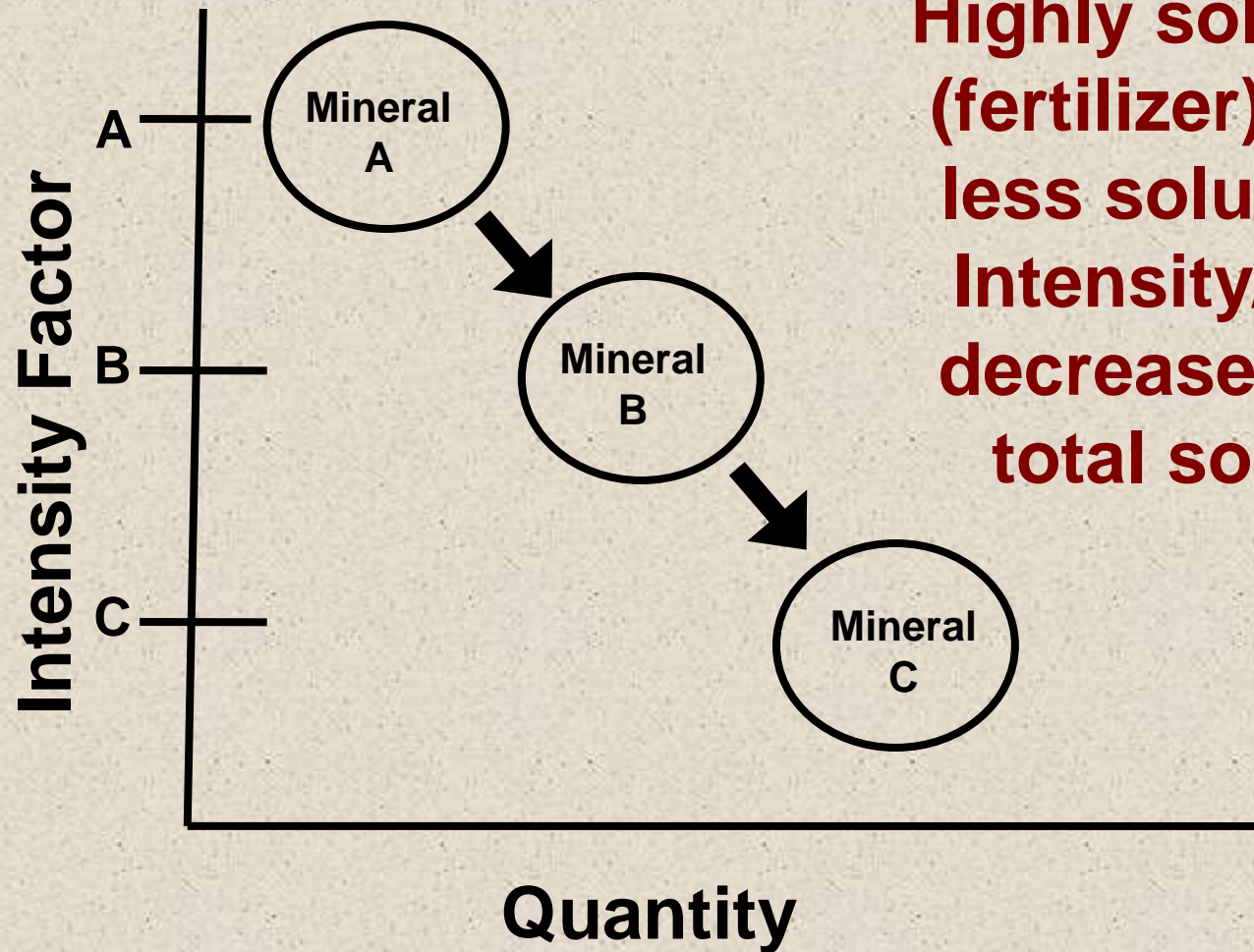
CaHPO_4 is metastable

Eventually transforms to FePO_4 , AlPO_4 or a Ca apatite

while CaHPO_4 is still around--
enjoy a higher soluble P and
plants grow well

soluble P fertilizer works

Effect of Q/I on P Availability / Solubility



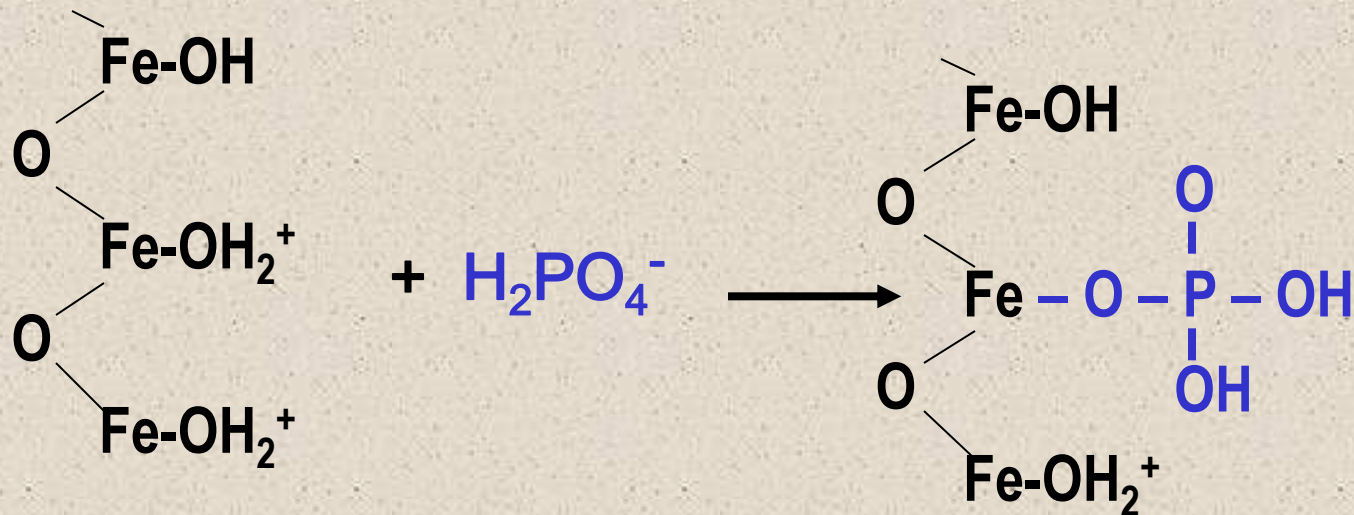
Highly soluble P minerals (fertilizer) transform into less soluble P minerals. Intensity/concentration decreases and quantity/total soil P increases

Perhaps this is why runoff P is often not related to soil test P

P Adsorption (binding) onto Oxide Clay Surfaces

Second mechanism controlling soil P solubility, mobility and availability

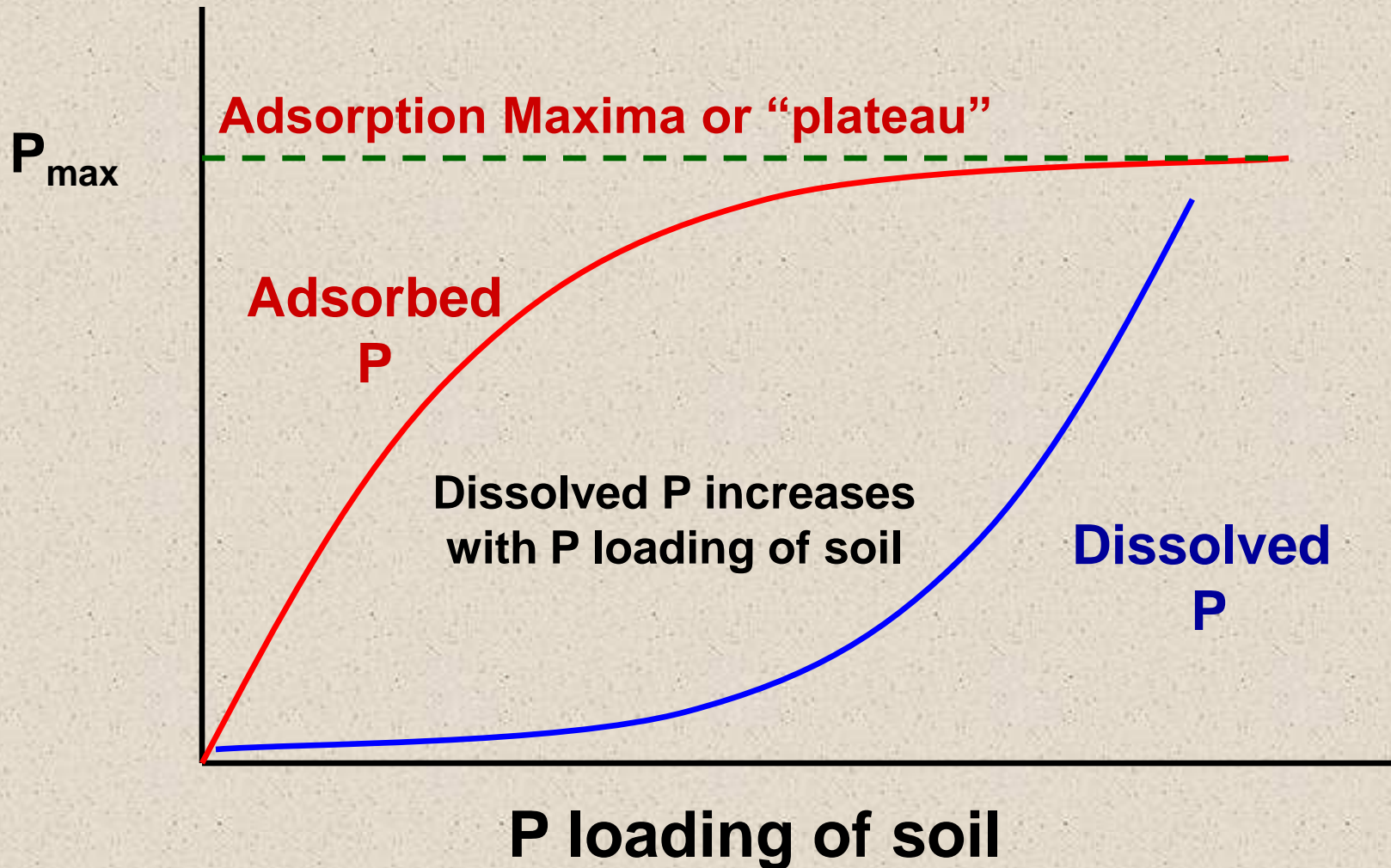
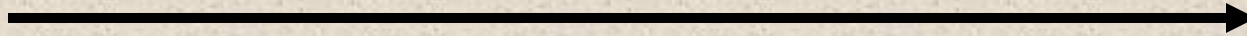
P adsorbs strongly to Al, Fe, Mn oxide soil surfaces



- **Dependent on reactive oxide surface availability**
- **Not concentration dependent like mineral precipitation/dissolution reactions**

Dissolved vs Adsorbed P

Increasing Phosphorus Loading of Soil



Risk of P Transport in Ohio

In Ohio, the risk of agricultural phosphorus (P) runoff transport into surface water is assessed by either the Ohio USDA-NRCS

Phosphorus Index (P Index) Assessment Procedure
or the
Soil Test Risk Assessment Procedure (STRAP)

within the
Nitrogen and Phosphorus Risk Assessment Procedures

[http://efotg.nrcs.usda.gov/references/public/OH/Nitrogen
and Phosphorous Risk Assessment Procedures.pdf](http://efotg.nrcs.usda.gov/references/public/OH/Nitrogen_and_Phosphorous_Risk_Assessment_Procedures.pdf)

Both are currently being reviewed

Predicting P solubility and risk

Ohio NRCS P Index

$$\begin{array}{l} \text{P Source Factors} \\ \text{Soil Test P} \\ \text{Soluble P In} \\ \text{Manure/biosolids} \end{array} + \begin{array}{l} \text{Transport Factors} \\ \text{Runoff Potential (slope)} \\ \text{Distance To Stream} \\ \text{Erosion Class, etc.} \end{array} = \text{P Index}$$

The Ohio P Index combines well established factors that influence the transport of P from agricultural fields to surface waters. Each factor is evaluated based on site specific data and weighted, or modified, according to its overall effect on P transport. Each of the site sub-values are added together to establish an overall site rating, or score, of Low, Moderate, High or Very High risk

Phosphorus source terms / weighting factors in current Ohio P Index

4	Soil Test P mg/kg	Bray-Kurtz-P1 mg/kg X 0.07				
5	Fertilizer Application Rate	Planned Fertilizer P₂O₅ Application (lbs/A) X 0.05				
6	Fertilizer P₂O₅ Application Method	0 Applied Value = 0	Immediate Incorporation OR Applied on 80% Cover Value = 0.75	Incorporation < 1 Week OR Applied on 50-80% Cover Value = 1.5	Incorporation > 1week and < 3 months OR Applied on 30- 49% Cover Value = 3.0	No Incorporation or Incorporation > 3 months OR Applied on < 30% Cover Value = 6.0
7	Organic P Application Rate	Planned Manure/Biosolids P₂O₅ Application (lbs/A) X 0.06				
8	Organic P₂O₅ Application Method	0 Applied Value = 0	Immediate Incorporation OR Applied on 80% Cover Value = 0.5	Incorporation < 1 Week OR Applied on 50-80% Cover Value = 1.0	Incorporation > 1week and < 3 months OR Applied on 30- 49% Cover Value = 2.0	No Incorporation or Incorporation > 3 months OR Applied on < 30% Cover Value = 4.0

The Soil Test Risk Assessment Procedure (STRAP)

**Simpler than P Index
Often used as a screening tool**

Used to predict risk of P transport based on the Bray-P1 extractable STP level. As STP levels increase above **150 mg/kg Bray-P1 it is presumed that there will be an increase in P transport and no additional phosphorus application is recommended.**

Once field STP levels exceed 150 mg/kg Bray-P, the P Index can be used to evaluate **risk of P transport and it is possible that the use of the P Index may allow for additional P application.**

Soil Test P (STP)

Important Component for BOTH Ohio Indices

Crop Sufficiency vs Environmental Risk

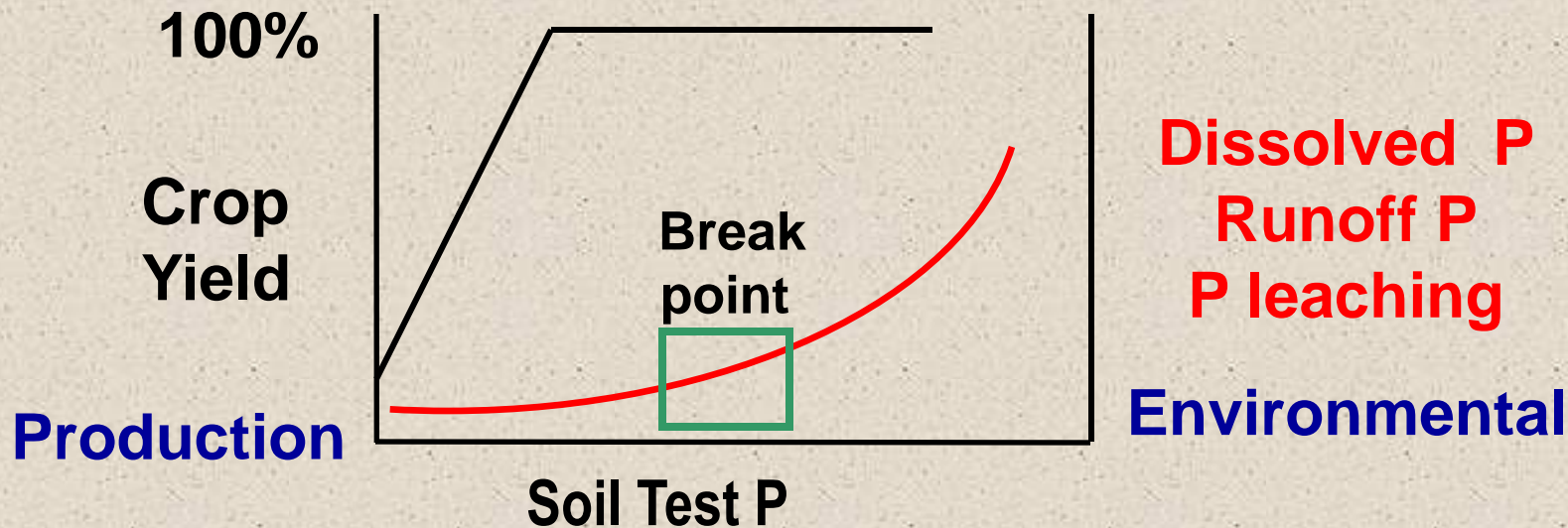
Crop production soil tests: Measure the “labile” nutrient pool that will be available to the crop during some / all of the growing season

Professor Bray’s Requirement for a “good soil test” (1948)

- 1. Extract a proportionate part of plant available forms of nutrient from different soils**
- 2. Extracted nutrients should correlate with uptake of nutrient and crop yield**

Is it appropriate to extrapolate soil fertility test data to predict risk of P transport to surface water (runoff/leaching)?

Using Crop Production Soil Tests as Environmental Soil Tests



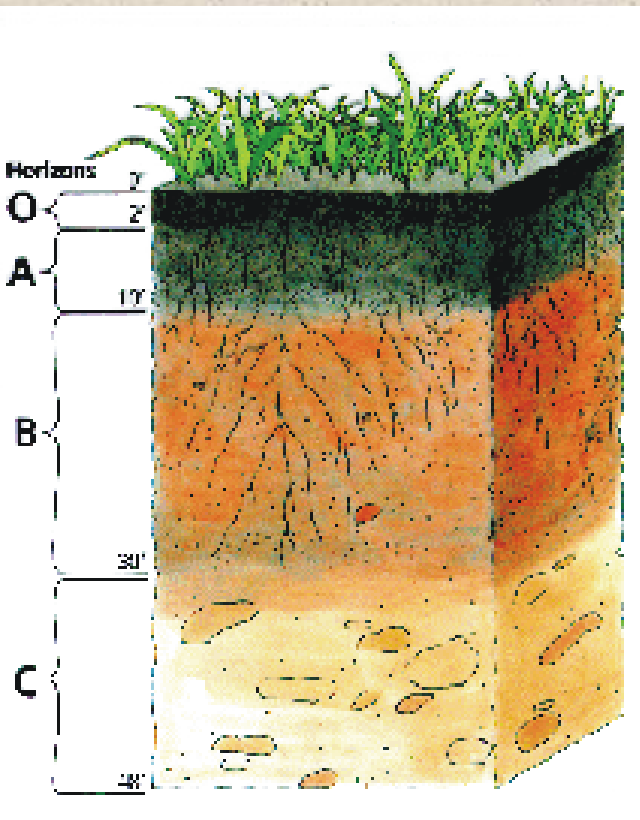
Identify soil test level “break point” where P solubility begins to increase rapidly **if there is one**

Set STP limit at break point to minimize soluble P

Problems: Are we confident where the break point is?
What depth is the soil sampled to?

Soil Depth and P Sorption

Typical Soil Profile



A Horizon

zone of organic matter accumulation
Typically less clay than B horizon
lower P sorption than B horizon

B Horizon

zone of clay (oxide) accumulation
increase in P adsorption
decrease in soluble P

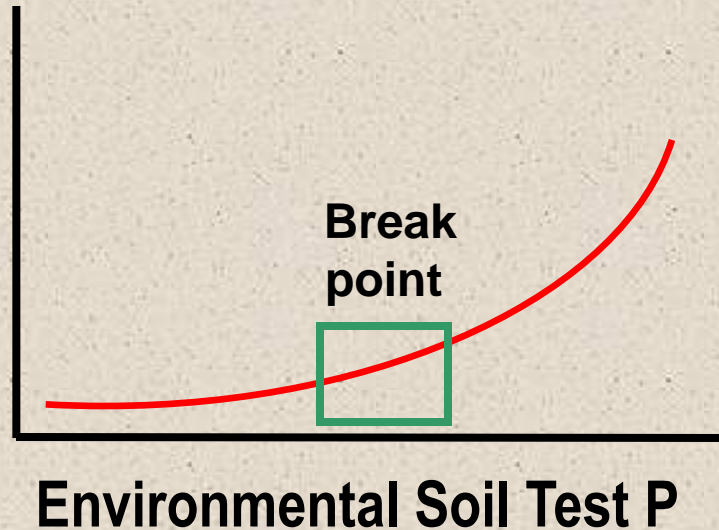
**Non conservation tillage:
P incorporation 6 to 8 inch depth**

**Surface fertilizer application or Conservation tillage:
incorporation of P (< 2 inch) more P in A horizon
which has lower sorption capacity.**

Potential for increased soluble P at the surface

Environmental P Soil Tests

Dissolved P
Runoff P
P leaching



Similar approach – evaluate other test methods not necessarily correlated to crop yield.

Evaluate surface (0-2 in) soil sampling vs agronomic soil sampling (0-8 in) to determine which (if either) best predicts risk of P transport (runoff / leaching)

Current and Future Research Goals

Evaluate (validate) and if necessary revise

- **P Index P source terms [Soil test P (STP), P application rate, and P application methods]**
- **Weighting/modifying factors to make P Risk Index more quantitative**
- **The threshold STP level of 150mg P/kg used in STRAP to predict P transport risk.**

Results could be directly applied toward revising Ohio P assessment tools as necessary.

**Long-term experiment at:
Waterman Research Farm
Columbus, OH
no-till, silt loam**



**CIG: Grand Lake St Mary's:
Mercer County
Celina, OH
no-till, silt loam**



**GLNPO-R4 Sandusky
Watershed
mixed tillage
silt loam and silty clay loam**

**GLNPO-R5, Robert Mullen
Maumee watershed
mixed tillage, silty clay loam**





Questions??

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